NOVEL D-GLUCOSE DERIVED HARD CARBON ANODE FOR SODIUM-ION BATTERIES

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The synthesis and characterization of a hard carbon anode material for room-temperature sodiumion battery is being reported. The carbon material was prepared via two-step process including hydrothermal carbonization (HTC) and pyrolysis. Similar carbon materials were recently synthesized and investigated by our group as supercapacitor electrode materials [1,2]. Sodium-ion batteries have emerged as a promising candidate for large-scale energy storage due to sodium's abundance and low cost of raw materials. However, significant work is needed on the development of Na-ion intercalation anodes [3,4].

The electrode slurry was prepared by mixing the active material, conductive additive (Super P) and polyvinylidene difluoride (PVdF) binder in 75:15:10 mass ratio. The mixed slurry was cast onto copper foil using doctor-blade technique. The cast electrodes were dried under vacuum for 24h. The half-cells (EL-Cell GmbH) were assembled and electrolyte mixtures prepared in an Argon-filled glovebox ($O_2 < 0.1$ ppm, $H_2O < 0.1$ ppm). Sodium metal was used as counter and reference electrode and 1M NaClO₄ in propylene carbonate (PC) as the electrolyte.

Acknowledgements

European Regional Development Fund: Centres of Excellence, 2014-2020.4.01.15-0011 and 3.2.0101–0030, TeRa project SLOKT12026T, Institutional Research Grant IUT20–13 and personal research grants PUT55 and PUT1033. The authors thank Prof. K. Kirsimäe for LA-ICP-MS analysis. Mr. Väli thanks Estonian students fund in USA and University of Tartu Foundation for financial support.

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