

# MAGNETIC PROPERTIES OF ATOMIC LAYER DEPOSITED MULTILAYER THIN SOLID FILMS

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Research devoted to the achievement of multiferroic thin films is active due to prospective applications as these materials would be useful as components of non-volatile memory and electromagnetic sensor devices. On the other hand, the task of depositing multiferroic thin films has appeared to be quite challenging and intensely ongoing. In relation to this,  $\text{Fe}_2\text{O}_3$ - $\text{BiOCl}$  composites [1], and  $\text{Co}_3\text{O}_4$ - $\text{ZrO}_2$  [2],  $\text{ZrO}_2$ - $\text{Al}_2\text{O}_3$  [3] and  $\text{HfO}_2$ - $\text{Al}_2\text{O}_3$  [4] nanolaminates were deposited and characterized in this study. The main focus was on studying the magnetic properties of these structures in their as-deposited states. Selected multilayer structures with their thicknesses and measured magnetic coercivity values are presented in Table I. The deposited structures demonstrated ferromagnetic-like behaviour at room temperature, and  $\epsilon$ - $\text{Fe}_2\text{O}_3$  containing specimens exhibited several times higher coercivity (reaching up to  $\sim 10$  kOe) compared to other samples. The stabilization of metastable phases induced magnetization in nanolaminates containing  $\text{ZrO}_2$  and  $\text{HfO}_2$ , which are oxides not known for their magnetic properties in their stable forms.

*Table I. Multilayer structure description (number in front of the compound indicates the number of ALD cycles), their total thicknesses and measured coercivity values*

Type of study	ALD structure description	Total thickness (nm)	Coercivity (Oe)
$\text{Fe}_2\text{O}_3$ - $\text{BiOCl}$ composites [1]	$175 \times \text{Fe}_2\text{O}_3 + 280 \times \text{BiOCl}$	50	9757
	$80 \times \text{Fe}_2\text{O}_3 + 280 \times \text{BiOCl}$	33	4230
$\text{Co}_3\text{O}_4$ - $\text{ZrO}_2$ [2]	$2 \times [200 \times \text{Co}_3\text{O}_4 + 100 \times \text{ZrO}_2] + 200 \times \text{Co}_3\text{O}_4$	64	32
	$2 \times [100 \times \text{ZrO}_2 + 200 \times \text{Co}_3\text{O}_4] + 100 \times \text{ZrO}_2$	60	21
$\text{ZrO}_2$ - $\text{Al}_2\text{O}_3$ [3]	$5 \times [120 \times \text{ZrO}_2 + 6 \times \text{Al}_2\text{O}_3] + 120 \times \text{ZrO}_2$	38	85
$\text{HfO}_2$ - $\text{Al}_2\text{O}_3$ [4]	$3 \times [200 \times \text{HfO}_2 + 10 \times \text{Al}_2\text{O}_3] + 200 \times \text{HfO}_2$	48	127

## References

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